

## Viscosity Activity Worksheet **Answers**

1. Describe the fluid you are working with using every day descriptive vocabulary. (For example: “I am looking at honey. It is yellow(ish) and clear(ish). It is pretty thick and moves slowly. It feels sticky.”)

Answers to questions 1-6 will vary, depending on the fluids and equipment used.

2. Calculate the density of the fluid using these steps:

- Weigh the empty graduated cylinder. Record its mass in grams.

$$M_{\text{cylinder}} = \text{_____} \text{ [g]}$$

- Fill the cylinder with fluid, and record the volume in  $\text{cm}^3$ . Note:  $1 \text{ cm}^3 = 1 \text{ ml}$ .

$$V_{\text{fluid}} = \text{_____} \text{ [cm}^3\text{]}$$

- Weigh the full graduated cylinder. Subtract the mass of the empty graduated cylinder and record the mass of the fluid.

$$M_{\text{fluid}} = \text{_____} \text{ [g]}$$

- The density of the fluid is the mass over the volume. Calculate the density of the fluid.

$$\rho_f = \frac{\text{Mass of Fluid [g]}}{\text{Volume of Fluid [cm}^3\text{]}}$$

$$\rho_f = \text{_____} \text{ [g/cm}^3\text{]}$$

Name: \_\_\_\_\_ Date: \_\_\_\_\_

3. Measure the density of the sphere using these steps:

- Measure the radius of the sphere. Record as r [cm].

$$r_s = \text{_____ [cm]}$$

- Calculate the volume of the sphere. Either use the equation:  
or place the sphere in a graduated cylinder filled with water and record its displacement.

$$Vol_s = \frac{4}{3}\pi r^3$$

$$Vol_s = \text{_____ [cm}^3\text{]}$$

- Weigh the sphere. Record its mass.

$$M_s = \text{_____ [g]}$$

- Calculate the density of the sphere by dividing its mass by its volume.

$$\rho_s = \frac{\text{Mass of Sphere [g]}}{\text{Volume of Sphere [cm}^3\text{]}}$$

$$\rho_s = \text{_____ [g/cm}^3\text{]}$$

4. Measure the terminal velocity of the sphere falling through the fluid using these steps:

- With your stopwatch ready, drop the ball into the fluid.

If the fluid is not very viscous, the ball will fall through it very fast, *so be ready!*

If the fluid is thick enough, then the ball will reach a constant speed.

This is the *terminal velocity*, the point at which the drag on the sphere by the fluid is equal to the force of gravity.

- Measure how fast the ball falls a distance. Record the distance, and the time.

$$\text{distance} = \text{_____ [cm]}$$

$$\text{time} = \text{_____ [s]}$$

- Calculate the velocity, which is the distance divided by the time.

$$V_s = \text{_____ [cm/s]}$$

Name: \_\_\_\_\_ Date: \_\_\_\_\_

**5. Using this equation, derived from Stokes' law, calculate the viscosity of your fluid.**

Gravity is  $981 \text{ cm/s}^2$ . *Be very careful* to show your units and how they cancel out.

Your final answer should be in units of  $[\text{g}/(\text{cm s})]$ .

$$\mu = \left(\frac{2}{9}\right) * \frac{r^2 * g(\rho_s - \rho_f)}{V_s}$$

$\mu =$  \_\_\_\_\_  $[\text{g}/(\text{cm s})]$

**6. Viscosities are usually recorded in  $[\text{Pa s}]$ . To convert from  $[\text{g}/(\text{cm s})]$  to  $[\text{Pa s}]$ , simply divide by 10:**

$$1[\text{Pa S}] = 1\left[\frac{\text{kg}}{\text{m s}}\right] = 1\left[\frac{1000 \text{ g}}{100 \text{ cm s}}\right] = 10\left[\frac{\text{g}}{\text{cm s}}\right]$$

$\mu =$  \_\_\_\_\_  $[\text{Pa s}]$

**7. Using the internet, look up the viscosities of some common household fluids.**

Be sure to include units. Do any of the answers surprise you?

Fluid	Viscosity	Fluid	Viscosity
<i>Example: blood</i>	$3 \times 10^{-3}$ to $4 \times 10^{-3}$ $[\text{Pa S}]$	castor oil	0.25 to 0.5
SAE 30 motor oil	0.25 to 0.5 $[\text{Pa S}]$	ketchup	50 to 70 $[\text{Pa S}]$
maple syrup	0.15 to 0.2 $[\text{Pa S}]$	shortening or lard	1,000 to 2,000 $[\text{Pa S}]$
milk	$2 \times 10^{-3}$ to $5 \times 10^{-3}$ $[\text{Pa S}]$	honey	2 to 3 $[\text{Pa S}]$

**Note:** In searching the internet, you may find viscosities in a variety of units. Some may be in Poise  $[\text{P}]$  or Centipoise  $[\text{cP}]$ .  $1 [\text{cP}] = 0.001 [\text{Pa s}]$ . The viscosity of water is  $1 [\text{cP}]$ . Other fluids may have viscosity in Stokes  $[\text{St}]$ , which is the ratio of the viscosity to the density of the fluid. To convert from Stokes, multiply it by the fluid's density, or find another source! *Hint:* Search for "dynamic viscosity."